<u>Statistical physics HW solutions</u> Solutions by Yury Kiselev

- 1. (a) Summing up all the heights and dividing by number of students, 10, gives 1.75.
 - (b) Histogram is plotted in Fig. 1b. See the end of the file.
 - (c) We will use middle values of height from the histogram to answer this question. The answer is $\langle h \rangle = \frac{1*1.55+2*1.65+3*1.75+3*1.85+1*1.95}{10} = 1.76$ m.
 - (d) From the histogram, it's 4 people out of 10, so 40%.
 - (e) 1/10, 2/10, 3/10, 3/10, 1/10 respectively.
 - (f) 1.0.
- 2. (a) It's 32.
 - (b) We will use middle values of velocities from the graph to answer this question. The answer is $\langle v \rangle = \frac{2*600+4*700+6*800+9*900+7*1000+3*1100+1*1300}{32} \approx 891 \text{ mi/h}$
 - (c) The answer is $\langle v^2 \rangle = \frac{2*600^2 + 4*700^2 + 6*800^2 + 9*900^2 + 7*1000^2 + 3*1100^2 + 1*1300^2}{32} \approx 817 \cdot 10^3 \text{ mi}^2/\text{h}^2$
 - (d) It's defined as a square root of the latter, so

$$v_{rms} = \sqrt{\langle v^2 \rangle} \approx 904 \text{ mi/h}$$

- (e) No, it's not the same, $\langle v^2 \rangle \neq \langle v \rangle^2$.
- (f) We will use half of one bin and half of the other, so answer is $p = (4/2+6/2)/32 = 5/32 \approx 16\%$.
- (g) No atoms here, so zero.
- (h) 2/32, 4/32, 6/32, 9/32, 7/32, 3/32, 1/32 respectively.
- 3. (a) The distance from the axis to the atoms is l/2, so $I = 2m(l/2)^2 = ml^2/2$.
 - (b) The distance from the axis to the atoms is zero in this case, so I = 0.
 - (c) $K_1 = \frac{1}{2}I_1\omega^2 = \frac{1}{2}I_1(2\pi)^2 f^2 = \pi^2 m l^2 f^2.$
 - (d) $K_2 = 0$, because $I_2 = 0$.
- 4. Gases are in equilibrium, so they have same temperatures. $v_{rms} = \sqrt{\frac{3kT}{m}}$, so the ratio of the velocities is inversely proportional to the square root of the ratio of the mass of the molecules. Helium is a monoatomic gas, while hydrogen is diatomic, so

$$\frac{v_{He}}{v_{H2}} = \sqrt{\frac{m_{H2}}{m_{He}}} = \sqrt{\frac{2}{4}} = \frac{1}{\sqrt{2}}$$

- 5. (a) $\langle v_z \rangle \propto \int_{-\infty}^{\infty} v_x \cdot e^{-mv_x^2/kT} dv_x = 0$, because this integral is an odd function.
 - (b) Boltzmann factors for other components will be different, $e^{-mv_x^2/kT}$ and $e^{-mv_y^2/kT}$ respectively, but the integral will look the same and will result in zero as well.
 - (c) No, because $\langle v^2 \rangle = \langle v_x^2 \rangle + \langle v_y^2 \rangle + \langle v_z^2 \rangle \neq \langle v_x \rangle^2 + \langle v_y \rangle^2 + \langle v_z \rangle^2$.
- 6. (a) Assuming that the vertical axis is z as usual, there is no energy associated with the difference in y coordinates, so the Boltzmann factor equals to 1. The average position is then $\langle y \rangle \propto \int_{y_{min}}^{y_{max}} 1 \cdot dy = 0.$
 - (b) This average is given by $\langle \vec{r} \rangle = (\langle x \rangle, \langle y \rangle, \langle z \rangle)$. I will assume that the container is small (not comparable to the size of the atmosphere), so the Boltzmann factors for all coordinates are equal to 1 (as $mgz \ll kT$). With this assumption, $\langle \vec{r} \rangle = (0, 0, 0)$ center of the cube.

